

Alteration in Growth and Physiological Activities in *Chlorella vulgaris* under the Effect of Photosynthetic Inhibitor Diuron

KHALAF ALI FAYEZ¹ AND ZEINAB ABD-ELFATTAH

Botany Department, Faculty of Science, Sohag University, 82524-Sohag, Egypt

¹Corresponding author e-mail: khalaffayez@yahoo.com

ABSTRACT

In this study, we describe the effect of diuron, a photosynthesis-inhibiting herbicide, on growth, photosynthetic pigment contents, carbohydrate, protein, amino acids, proline, nitrate reductase and some antioxidant enzymes in cell suspensions of *Chlorella vulgaris* for seven days. The growth (cell number & dry weight) was seriously affected even with lowest diuron dose (0.1 μM). Photosynthetic pigment contents (chl-a, chl-b & caroteniod) of diuron-treated algae, was more sharply decreased as compared to control. Diuron induced oxidative stress as indicated by the alteration of ascorbate oxidase. Nitrate reductase (NR) activity in the volume of algal suspension was significantly decreased with increasing diuron doses. Guaiacol peroxidase was un-detectable in diuron-treated and un-treated, which may be due to hypersensitivity of *C. vulgaris* towards toxic compounds. Proline, total amino acids, carbohydrate and protein contents of algal suspension decreased significantly in response to diuron treatments. All results showed a negative correlation with treatments except ascorbate oxidase activity.

Key Words: Antioxidant enzymes; *C. vulgaris*; Diuron; Growth; Nitrate reductase; Pigments

INTRODUCTION

Herbicides are the principal method of weed control but the introduction of these compounds into the aquatic environment through runoff may have severe consequences for non-target plants. The widespread use of herbicidal chemicals in weed control has created problems of toxic residues in the natural environment (Glottelty *et al.*, 1984; Leonard, 1990). The adverse effects of these pollutants on non-target plants are particularly of concern due to their increasingly worldwide use (Van den Brink & Ter Braak, 1999). Phenylurea herbicides are widely used throughout the world for the protection of several crops and consequently, can have residual effect on crops, soil and surface waters (Peña *et al.*, 2002).

Diuron, 3-(3,4-dichlorophenyl)-1,1-dimethylurea, has been shown to inhibit photosystem-II (PSII) electron transport, chlorophyll content and affect growth, cell organelles and metabolism of photosynthetic plants (Oettmeier, 1999; Fayez, 2000a; Eullaffroy & Vernet, 2003). Moreover, the herbicide diuron is commonly incorporated into antifouling paints to boost the efficacy of the compound towards algae (Chesworth *et al.*, 2004).

Algae play an important role in the equilibrium of aquatic ecosystems, being the first level of the trophic chain to produce organics and oxygen (Campanella *et al.*, 2000). The use of photosynthetic organisms in toxicity tests is very appropriate because more than 65% pesticides are herbicides (USDA, 1998). Furthermore, about 50% of these herbicides act via inhibition of photosynthesis at the photosystem-I (PSI) and PSII levels by replacing PSI's ultimate electron acceptor (Powles *et al.*, 1997) or by blocking PSII-catalyzed photosynthetic electron transport.

Pigments have often been used as biomarkers of exposure to different class of herbicides in autotrophic plants including algae (Blaise, 1993; Sandmann, 1993).

In this study, we investigated the response of *Chlorella vulgaris* to diuron (PS II inhibitor) in terms of growth, photosynthetic pigment contents, activity of nitrate reductase, some antioxidant enzymes and metabolites of carbohydrate, protein and amino acids were determined.

MATERIALS AND METHODS

Algal culture conditions. The freshwater unicellular green algae *C. vulgaris* was isolated from Nile water. An axenic culture was maintained in 500 mL glass Erlenmeyer flasks filled with 250 mL of Beijerinck medium (Stein, 1966). The pH was adjusted at 6.8 before sterilization. Cultures were maintained at 25°C under a light intensity of 160 $\mu\text{mol m}^{-2} \text{s}^{-1}$ with 16 h light period, shaken at 60 rpm on an orbital shaker and kept in suspension by bubbling air. Initial density of the *C. vulgaris* cultures was $273 \times 10^4 \text{ cells mL}^{-1}$ (Fig. 1). The optimal pigment (chl- a, b & caroteniod) contents for algal suspension experiments, was found to be 3.03 $\mu\text{g mL}^{-1}$. Measurements were performed at zero time (prior to algal treated with herbicide).

Application of herbicide. Diuron herbicide stock solutions were made in acetone followed by 0.1 – 0.5 μM concentrations prepared from stock. Algal solutions containing equivalent solvent concentrations in the absence of herbicide were used as controls. All experiments were replicated three times.

Cell number and its dry weight. Hemacytometer, 0.1 mm deep, having improved Naubauer ruling (A.O. Spencer "Bright fine") was used. The mean counts of three replicates were taken in consideration and the data were given as cell

mL⁻¹ algal suspension. Dry weight was determined according to Utting (1985). Culture aliquots (50 mL) were filtered through previously dried and weighed Whatman GF/C filters. Thereafter the filters were dried in an oven at 80°C for 72 h.

Photosynthetic pigment extraction. Chlorophyll *a*, *b* and Carotenoid were extracted in 100% methanol at 65°C and their contents were determined spectrophotometrically (Spekol 11, Carl Zeiss, Jena, Germany) according to Metzner *et al.* (1965).

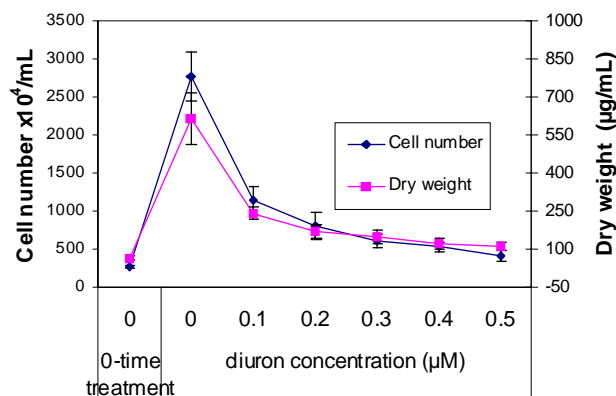
Nitrate reductase assay *in vivo*. For *in vivo* assay of nitrate reductase (EC 1.7.1.1), the method of Jaworski (1971) was used. Algal cells of 10 mL algal suspension of diuron-treated alga and un-treated were precipitated and incubated anaerobically in dark for 1 h in 5 mL of 0.1 M K-phosphate (pH = 7.5) containing 50 mM KNO₃ and 1% (v/v) n-propanol at 28°C. The reaction was stopped by boiling in water bath for 5 min and then centrifuged. The supernatant of one mL sample mixed well with two mL 1% w/v sulphonilamide in 1 N HCl and 2 mL 0.1% w/v N-(1-naphthyl) ethylenediamine dihydrochloride in distilled water. The absorbency was measured by using spectrophotometer at 540 nm. Nitrate reductase activity was expressed as ng NO₂ mL⁻¹ algal suspension h⁻¹.

Ascorbic acid oxidase (EC 1.10.3.3) and peroxidase (EC 1.11.1.7) determination. For ascorbic acid oxidase, the method of Olliver (1967) modified by Chinoy (1976) was used. The reaction mixture consisted of 1 mL of 0.1 mM ascorbic acid, 1 mL of 0.5 mM phosphate buffer (pH 6.8) and 2 mL of enzyme extract. This reaction mixture was incubated at 37°C for 25 min. After incubation the 2, 6-dichlorophenol indophenol was added and the absorbance was taken at 620 nm. The peroxidase activity was measured by following the change of absorption at 470 nm due to guaiacoin oxidation. The activity was assayed for 1 min in a reaction solution (3 mL final volume) composed of 100 mM potassium phosphate buffer (pH 7.0), 20 mM guaiacoin, 10 mM H₂O₂ and 50 µL of crude extract (Polle *et al.*, 1994).

Estimation of proline. Free proline content of algal suspension was determined according to Bates *et al.* (1973). Ten mL of algal suspension was centrifuged and the alga extracted in five mL of aqueous 3% sulfosalicylic acid for 3 h. The extract was centrifuged at 1500 rpm for 10 min. and two mL of the supernatant were mixed with two mL of fresh acid ninhydrin solution and 2 mL glacial acetic acid in a test tube for 1 h at 100°C. The reaction was terminated in an ice bath and the mixture was extracted with four mL toluene. The extract was vigorously stirred for 20 s using a test tube stirrer. Therefore the chromophore-containing toluene was aspirated from the aqueous phase and its absorbance was measured at 520 nm. Proline was used as a standard.

Protein content. Protein content was determined according to Lowry *et al.* (1951). Ten mL algal suspension was extracted in distilled-water (soluble protein) and in NaOH (total protein) for two h at 90°C. The extract was centrifuged and the supernatants were pooled. The water-

Fig. 1. Cell number and dry weight contents of diuron treated and untreated *C. vulgaris* for 7 days as well as the initial cell number and dry weight at zero time treatment. Values are the mean of three replicates. Vertical bars are SD. Statistical significance of differences compared to control. The values of cell number and dry weight are significant at all diuron doses ($P > 0.01$; $r = -0.84$ & -0.81 for cell number and dry weight, respectively)



soluble protein was estimated by the Folin-phenol reagents spectrophotometrically using bovine serum albumin was used as standard.

Carbohydrate content estimation. Carbohydrate content was determined in aqueous (soluble carbohydrate) and in HCl solutions (total carbohydrate) with anthrone sulphuric acid reagent according to Fales (1951), using glucose as a standard. The blue green color developed was measured at the 620 nm using spectrophotometer.

Statistical analysis. All data are the mean of three replicates. Data were statistically analyzed by a one-way ANOVA. Means in individual experiments were tested for significance at $P < 0.01$. Correlation coefficient (r) of results was also performed.

RESULTS AND DISCUSSION

After 7 days, the growth of *C. vulgaris* towards increasing concentrations of diuron in terms of cell number and dry weight was significantly decreased (Fig. 1). The cell number increased by tenfold (2760.333×10^4) compared to the initial cell number. After seven day of treatment, the cell number at 0.1 µM and 0.5 µM diuron decreased by 2.4 and 6.7 fold, respectively ($P < 0.01$; $r = -0.84$). The dry weight also decreased significantly with increasing diuron doses (Fig. 1). The dry weight of *C. vulgaris* after seven days of 0.1 and 0.5 µM diuron treatment was lowered by 2.5 and 5.5 fold, respectively in comparison with the control ($P < 0.01$; $r = -0.81$), which is inline with previous reports (Shehata *et al.*, 1997; Rioboo *et al.*, 2002).

Photosynthetic pigments were significantly decreased ($P < 0.01$; $r = -0.87$) in response to increasing diuron treatments (Table I). After 7 days treatments, the total pigment contents (chl-*a*, chl-*b* & carotenoid) at lowest (0.1 µM) and highest (0.5 µM) doses indicated 45.60 and 14.09% respectively of the control (Table I). Pigments deficiency can be caused by photobleaching or by inhibition

Fig. 2. Proline and total amino acid contents of diuron treated and untreated *C. vulgaris*. Values are the mean of three replicates. Vertical bars are SD. Statistical significance of differences compared to control. The values of proline and amino acids are significant at all diuron doses ($P > 0.01$; $r = -0.90$ & -0.75 for proline and total amino acids, respectively)

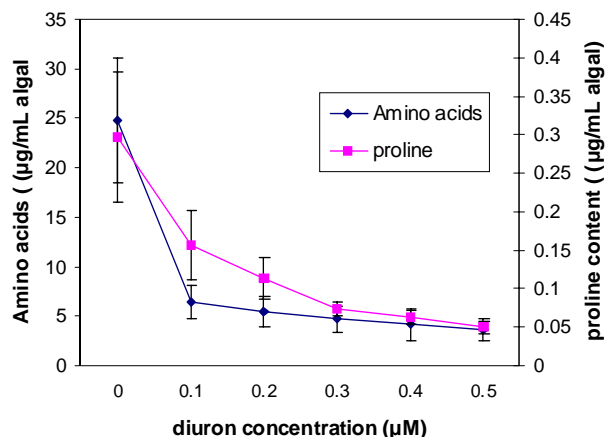
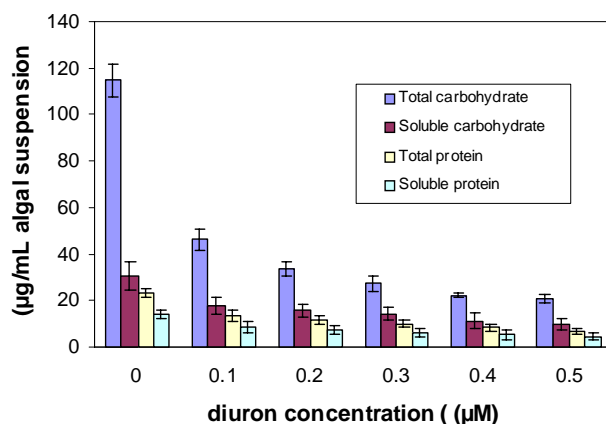


Fig. 3. Total and soluble metabolites of protein and carbohydrates of diuron treated and untreated *Chlorella vulgaris*. Values are mean of three replicates. Vertical bars are SD. Statistical significance of differences compared to control: The values of carbohydrates and proteins are significant at all diuron doses ($P > 0.01$; $r = -0.82$, -0.89 , -0.90 , -0.91 for total carbohydrate, soluble carbohydrate, total protein & soluble protein, respectively)



of their biosynthesis (Barry *et al.*, 1990; Faye, 2000a; Couderchet & Vernet, 2003). Böger and Sandman (1990) reported that photosynthetic inhibitors inhibited the formation of protoporphyrin IX in *Scenedesmus*, soybean and *Lemna*. Photosynthesis inhibition is an important indicator of the toxic effect of pollutants (Gonen-Zurgil, 1997; Hargers *et al.*, 1998). Ascorbate oxidase activity (Table II) was significantly enhanced in response to diuron doses comparing with the control ($P < 0.01$; $r = 0.67$). The highest activity (229% of control) was detected with 0.3 μM diuron. Such an effect might be due to blocking of photosynthetic electron transport (Barry *et al.*, 1990; Sandmann, 1993). Guaiacil peroxidase was un-detectable in diuron-treated and un-treated *C. vulgaris* (Table II). The absence of

Table I. Photosynthetic pigments (μg mL⁻¹ algal suspension) of diuron treated and untreated of *C. vulgaris* for 7 days. Values are the mean ± SD of three replicates. Statistical significance of differences compared to control: *, significant at $P > 0.01$; $r = -0.86$, -0.89 , and -0.86 for Chl-a, Chl-b and Carotenoid, respectively

Diuron doses (μM)	Chl-a Mean ± SD	Chl-b Mean ± SD	Carotenoid Mean ± SD	Total pigments	% of control
0.0	7.45 ± 0.88	2.88 ± 0.38	2.87 ± 0.25	13.2	100
0.1	3.32* ± 0.33	1.42* ± 0.31	1.27* ± 0.19	6.02	45.60
0.2	2.25* ± 0.27	1.05* ± 0.17	0.96* ± 0.25	4.26	32.27
0.3	1.81* ± 0.21	0.79* ± 0.11	0.78* ± 0.18	3.39	25.68
0.4	1.41* ± 0.47	0.55* ± 0.15	0.60* ± 0.24	2.57	19.47
0.5	0.98* ± 0.26	0.42* ± 0.12	0.46* ± 0.14	1.86	14.09

Table II. Enzyme activities of diuron treated and untreated *C. vulgaris* for 7 days. Values are the mean ± SD of three replicates. Statistical significance of differences compared to control: *, significant at $P > 0.01$; $r = 0.67$ and -0.89 for ascorbate oxidase and nitrate reductase activities, respectively

Diuron doses (μM)	Ascorbate activity (%)	oxidase NR activity of control ng NO ₂ mL ⁻¹ h ⁻¹	Guaiacil peroxidase activity
0	100	26.33 ± 2.08	Nd
0.1	121	20.33* ± 2.08	Nd
0.2	191	05.33* ± 1.52	Nd
0.3	229	04.40* ± 1.80	Nd
0.4	217	02.86* ± 0.40	Nd
0.5	160	02.26* ± 0.41	Nd

Nd = Non- detectable

guaiacil peroxidase may explain the hypersensitivity of the alga to diuron. Peroxidase is one of antioxidant enzymes scavenge reactive oxygen species (ROS). Oxidative stress occurs when a plant is subjected to biotic and abiotic stresses due to production of ROS (Mittler, 2002; Blokhina *et al.*, 2002; Clarke *et al.*, 2002; Ernani *et al.*, 2003).

In-vivo NR activity was significantly decreased in response to diuron treatments (Table II). The inhibition was most obvious with increasing diuron doses ($P < 0.01$; $r = -0.89$). At 0.1 and 0.5 μM diuron doses, the activity of NR was 77 and 8.6%, respectively of the control value. NR activity of plant was inhibited by various stresses (Aslam *et al.*, 1984; Megharaj *et al.*, 1993; Faye, 2000b; Quaggiotti *et al.*, 2004; Mallick, 2004).

Proline and total amino acids contents mL⁻¹ of algal suspension were significantly decreased (Proline: $P < 0.01$; $r = -0.89$ & total amino acids: $P < 0.01$; $r = -0.74$) with increasing diuron doses to those of the control (Fig. 2). One mL of algal suspension treated with lowest (0.1 μM) and highest dose (0.5 μM) of diuron was 52 and 17%, of control, respectively. Total amino acids contents mL⁻¹ of algal suspension at 0.1 and 0.5 μM of diuron were 25 and 15% of the control, respectively. However proline and amino acids contents expressed on cell dry weight basis showed an increase at various diuron doses. For example, the proline content of control and 0.1 μM diuron was 481 and 647 μg g⁻¹ dry weight of alga, respectively. Such an increase in proline level in plant roots has been reported, indicating modulated metabolic activities of cell under stress conditions (Faye & Kristen, 1996; Faye, 2000a). Protein

and carbohydrate contents (total & soluble) of algal also decreased significantly ($P < 0.01$) with increasing diuron doses (Fig. 3). However calculation the concentration of protein and carbohydrate on the basis of dry weight showed an increase in their contents with increasing diuron doses. The soluble protein content at control and lowest dose (0.1 μM) was 23 and 35 mg g^{-1} dry weight, respectively.

We conclude that changes in growth, pigment levels and metabolite constituents and antioxidants in *C. vulgaris* can be used effectively as tools to evaluate the toxicity effects of diuron and possibly of other photosynthetic inhibitors.

REFERENCES

- Aslam, M., R.C. Huffaker and D.W. Rains, 1984. Early effects of salinity assimilation in barley seedlings. *Pl. Physiol.*, 76: 321–5
- Barry, P., A.J. Young and G. Britton, 1990. Photodestruction of Pigments in Higher Plants by Herbicide Action. 1. The Effect of DCMU (diuron) on Isolated Chloroplasts. *J. Exp. Bot.*, 41: 123–9
- Bates, L.S., R.P. Waldren and I.D. Teare, 1973. Rapid determination of free proline for water-stress studies. *Pl. Soil*, 39: 205–7
- Blaise, C.R., 1993. Practical laboratory applications with micro-algae for hazard assessment of aquatic contaminants. In: Richardson, M. (ed.), *Ecotoxicology Monitoring*, pp: 83–107. VCH, Weinheim
- Blokhina, O., E. Virolainen and K.V. Fagersted, 2002. Antioxidants, Oxidative Damage and Oxygen Deprivation Stress: a Review. *Annu. Bot.*, 91: 179–94
- Böger, P. and G. Sandmann, 1990. Modern Herbicides Affecting Typical Plant processes. In: Bowers, W.S., W. Ebing, D. Martin and R. Wegler (eds.), *Chemistry of Plant Protection Vol. 6: Controlled Release, Biochemical effects of Pesticides, Inhibition of Plant Pathogenic Fungi*, pp: 174–210. Springer Publication, Berlin, Heidelberg
- Campanella, L., F. Cubadda, M.P. Sammartino and A. Saoncella, 2000. An algal biosensor for the monitoring of water toxicity in estuarine environments. *Water Res.*, 25: 69–76
- Chesworth, J.C., M.E. Donkin and M.T. Brownb, 2004. The interactive effects of the antifouling herbicides Irgarol 1051 and Diuron on the seagrass *Zostera marina* (L.). *Aquatic Toxicol.*, 66: 293–305
- Chinoy, J.J., Y.D. Singh and K. Gurumurthi, 1976. Colorimetric determination of ascorbic acid turnover in plants. *Indian J. Pl. Physiol.*, 19: 122–30
- Clarke, S.F., P.L. Guy, D.J. Burritt and P.E. Jameson, 2002. Changes in the activities of antioxidant enzymes in response to virus infection and hormone treatment. *Physiol. Pl.*, 114: 157–64
- Couderechet, M. and G. Vernet, 2003. Pigments as biomarkers of exposure to the vineyard herbicide flazasulfuron in freshwater algae. *Ecotoxicol. Environ. Safety*, 55: 271–7
- Ermani, P., C.S.S.K. Teresa, A.S.L. Maria, K.O. Oswaldo, M. David and C. Pio, 2003. Heavy metal-induced oxidative stress in algae. *J. Phycol.*, 39: 1008–18
- Eullaffroy, P. and G. Vernet, 2003. The F684/F735 chlorophyll fluorescence ratio: a potential tool for rapid detection and determination of herbicide phytotoxicity in algae. *Water Res.*, 37: 1983–90
- Fales, F.W., 1951. The assimilation and degradation of carbohydrates by yeast cells. *J. Biol. Chem.*, 193: 113–24
- Fayez, K.A. and U. Kristen, 1996. The influence of herbicides on the growth and proline content of primary roots and on the ultrastructure of root caps. *Environ. Exp. Bot.*, 36: 71–81
- Fayez, K.A., 2000a. Action of Photosynthetic Diuron Herbicide on Cell Organelles and Biochemical Constituents of the Leaves of Two Soybean Cultivars. *Pestic. Biochem. Physiol.*, 66: 105–15
- Fayez, K.A., 2000b. Response of physiological activities and cell ultrastructure of *lupinus termis* to saline water treatment. *Bull. Fac. Sci. Assiut University*, 29: 271–83
- Glottfelty, D.E., A.W. Taylor, A.R. Isensee, J. Jersey and S. Glenn, 1984. Atrazine and simazine movement to Wye river estuary. *J. Environ. Qual.*, 13: 115–21
- Gonen-Zurgil, Y., Y. Carmeli-Schwartz and A. Sukenik, 1997. Selective effect of the herbicide DCMU on unicellular algae – a potential tool to maintain monoalgal mass culture of *Nannochloropsis*. *J. Applied Phycol.*, 8: 415–9
- Hargers, E.M., G.H. (Ren'e) Aalderink, P.J. Van Den Brink, R.G.J. Wilfred, F. Wiegman and T.C.M. Brock, 1998. Ecotoxicological threshold levels of a mixture of herbicides (atrazine, diuron & metolachlor) in freshwater microcosms. *Aquatic Ecol.*, 32: 135–52
- Jaworski, E.G., 1971. Nitrate reductase assay in intact plant tissues, *Biochem. Biophys. Res. Commun.*, 43: 1274–9
- Leonard, R.A., 1990. Movement of pesticides into surface waters. In: Cheng, H.H. (ed.), *Pesticides in the Soil Environment: Processes, Impacts and Modeling*, pp: 303–49. Soil Science Society of America Inc, Madison, WI
- Lowry, O.H., N.S. Rosebrough, A.L. Farr and R.J. Randall, 1951. Protein measurement with the Folin-phenol reagent, *J. Biol. Chem.*, 193: 265–75
- Mallick, N., 2004. Copper-induced oxidative stress in the chlorophycean microalga *Chlorella vulgaris*: response of the antioxidant system. *J. Pl. Physiol.*, 161: 591–7
- Megharaj, M., H.W. Pearson and K. Venkateswarlu, 1993. Toxicity of carbofuran to soil isolates of *Chlorella vulgaris*, *Nostic linkia* and *N. muscorum*. *Appl. Microbiol. Biotechnol.*, 39: 644–8
- Metzner, H., H. Rau and H. Singer, 1965. Untersuchungen zur Synchronisierbarkeit einzelner Pigmentmangel Mutanten von *Chlorella*. *Planta*, 65: 186–94
- Mittler, R., 2002. Oxidative stress, antioxidants and stress tolerance. *Trends Pl. Sci.*, 7: 405–10
- Oettmeier, W., 1999. Herbicide resistance and super sensitivity in photosystem II. *CMLS Cell. Mol. Life Sci.*, 55: 1255–77
- Olliver, M., 1967. Ascorbic acid: Estimation. In: Sebrell, Jr W.H. and R.S. Harris (eds.), *The Vitamins*, Vol. I. 2nd edition, pp: 338–59. Academic Press, New York and London
- Peña, F., S. Cárdenas, M. Gallego and M. Valcárcel, 2002. Analysis of phenylurea herbicides from plants by GC/MS. *Talanta*, 56: 727–34
- Polle, A., T. Otter and F. Seifert, 1994. Apoplastic peroxidases and lignification in needles of Norway Spruce *Picea abies* L. *Pl. Physiol.*, 106: 53–60
- Powles, S.B., C. Preston, B. Bryanl and A.R. Jutsum, 1997. Herbicide resistance: impact and management. *Adv. Agron.*, 58: 57–93
- Quaggiotti, S., A.R. Trentin, F.D. Vecchia and R. Ghisi, 2004. Response of maize (*Zea mays* L.) nitrate reductase to UV-B radiation. *Pl. Sci.*, 167: 107–16
- Rioboo, C., O. González, C. Herrero and A. Cid, 2002. Physiological response of freshwater microalga (*Chlorella vulgaris*) to triazine and phenylurea herbicides. *Aquatic Toxicol.*, 59: 225–35
- Sandmann, G., 1993. Spectral determination of carotenoid precursors in *Scenedesmus* cells treated with bleaching herbicides. In: Böger, P., G. Sandmann (eds.), *Target Assays for Modern Herbicides and Related Compounds*, pp: 3–8. Lewis Publishers, F.L
- Shehata, S.A., M.A. El-Dib and H.F. Abou Waly, 1997. Effect of certain herbicides on growth of fresh water algae. *Water Air and Soil Pollution*, 100: 1–12
- Stein, J.R., 1966. Growth and mating of *Gonium pectorate* (Volvocales in defined media). *J. Phycol.*, 2: 23–8
- USDA, 1998. *Agricultural Chemical Usage: Field Crop Summary*. U.S Department of Agriculture, July 99, Washington, D.C, 1999
- Utting, S.D., 1985. Influence of nitrogen availability on the biochemical composition of three unicellular marine algae of commercial importance. *Aqua-cult. Eng.*, 4: 175–90
- Van Den Brink, P.J. and C.J.F. Ter Braak, 1999. Principal response curves: analysis of time-dependent multivariate responses of biological community to stress. *Environ. Toxicol. Chem.*, 18: 138–48

(Received 19 May 2007; Accepted 19 June 2007)